

A Review on Current Trends in High Performance Liquid Chromatography

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Abstract: *Chromatography is the backbone of separation science and is extensively applied in research laboratories and pharmaceutical industries worldwide. Among various chromatographic techniques, High-Performance Liquid Chromatography (HPLC) is the most widely used analytical tool due to its high sensitivity, accuracy, and versatility. Current trends in HPLC focus on achieving higher separation efficiency, improved resolution, and reduced analysis time. This article presents a concise review of HPLC, including its fundamental principles and instrumentation. It highlights recent advancements such as Rapid Resolution Liquid Chromatography (RRLC), Ultra-Performance Liquid Chromatography (UPLC), Ultra-Fast Liquid Chromatography (UFLC), and Nano Liquid Chromatography (Nano LC). A detailed comparison of these advanced techniques is provided with respect to instrumental operating parameters including column temperature, flow rate, injection volume, and pressure requirements. The article also discusses the applications and advantages of each technique over conventional HPLC. Recent innovations integrating advanced column technologies, mass spectrometric detection, automation, and green analytical approaches are emphasized, demonstrating the evolving role of HPLC in pharmaceuticals, environmental analysis, and clinical diagnostics.*

Keywords: Liquid chromatography, HPLC, RRLC, UPLC, UFLC, Nano LC

I. INTRODUCTION

High-Performance Liquid Chromatography (HPLC) is a powerful and widely used analytical technique that plays a vital role in almost every scientific discipline. It is primarily employed for the separation, identification, and quantification of compounds present in complex mixtures. Due to its high resolution, accuracy, sensitivity, and flexibility, HPLC has become indispensable in fields such as pharmaceuticals, environmental science, food safety, and biochemistry. The basic principle of HPLC involves passing a liquid sample through a column packed with a solid stationary phase. Different components of the sample interact with the stationary phase to varying degrees, leading to their separation based on affinity. Factors such as the nature of the mobile phase, stationary phase, column temperature, and detection method strongly influence this process, and proper optimization of these parameters is essential for successful method development.

The primary objective of HPLC method development is to establish optimal analytical conditions that ensure accurate, precise, and reproducible results for specific analytes. This involves selecting suitable solvents, pH conditions, gradient or isocratic elution modes, column types, and detection techniques. Sample preparation methods, such as filtration or extraction, are also crucial to improve analyte recovery and minimize matrix interference. Once a method is developed, it must undergo rigorous validation to confirm its suitability for the intended purpose. Validation parameters include accuracy, precision, specificity, sensitivity, linearity, range, robustness, and system suitability, in accordance with regulatory guidelines such as ICH and USP.

Over time, HPLC technology has advanced significantly. Modern developments such as Ultra-High-Performance Liquid Chromatography (UHPLC) and hyphenated techniques like HPLC-MS/MS have greatly enhanced sensitivity, reduced analysis time, and improved detection limits. These advancements have strengthened the role of HPLC as a cornerstone



technique in pharmaceutical analysis, supporting drug discovery, formulation studies, impurity profiling, pharmacokinetic analysis, quality control, and regulatory compliance. Recent innovations in column technologies, including sub-2 μm particles, core-shell particles, and monolithic columns, have improved separation efficiency and reduced solvent consumption. Optimization of mobile phase composition, including the use of buffers, alternative solvents, and environmentally friendly additives, has further enhanced chromatographic performance. Advances in detection systems such as UV-Vis, fluorescence, and mass spectrometry have increased sensitivity and selectivity, allowing trace-level analysis in complex matrices.

New Trends in HPLC Techniques

Modern advancements in High-Performance Liquid Chromatography (HPLC) have focused on achieving faster analysis, higher resolution, and improved sensitivity. Compared with classical HPLC, newer chromatographic techniques include Rapid Resolution Liquid Chromatography (RRLC), Ultra-Performance Liquid Chromatography (UPLC), Ultra-Fast Liquid Chromatography (UFLC), and Nano Liquid Chromatography (Nano LC). These techniques are designed to meet the increasing analytical demands of pharmaceutical, environmental, and biochemical research.

Rapid Resolution Liquid Chromatography (RRLC)

Rapid Resolution Liquid Chromatography (RRLC) is designed to deliver very high analysis speed, resolution, and pressure with maximum efficiency. It is one of the fastest and most flexible liquid chromatography systems available and has become a routine analytical method in the pharmaceutical industry. RRLC offers excellent peak shapes, enhanced reproducibility, high sensitivity, and high-speed detection while reducing overall analysis cost. These advantages make it especially useful for quality control applications, including the analysis of herbal medicines.

The reduction in analysis time in RRLC is mainly achieved by using shorter columns. However, shorter columns can reduce the number of theoretical plates and compromise resolution. To overcome this limitation, RRLC employs columns packed with sub-2 μm particles, which significantly increase column efficiency. The use of smaller particle sizes allows high resolution to be maintained even with shorter columns, enabling rapid analysis without loss of separation quality. Higher flow rates and elevated column temperatures (up to 100 $^{\circ}\text{C}$ for certain columns) are often applied to reduce system back-pressure and further enhance performance.

RRLC can theoretically provide up to 20-fold faster analysis compared to conventional HPLC while maintaining equivalent resolution. It improves sensitivity, supports true fast LC analysis, and offers greater selectivity flexibility. These features result in increased sample throughput and reduced solvent and operational costs.

RRLC has wide applications in pharmaceutical and environmental analysis. It is used in RRLC-MS/MS methods for the determination of endocrine-disrupting chemicals, pharmaceuticals, and personal care products in environmental samples. It is also applied in quality control of herbal products such as *Rhodiola rosea*, *Panax*, and *Epimedium* species. Additionally, RRLC is used for impurity profiling, polycyclic aromatic hydrocarbon (PAH) separations, benzodiazepine analysis, and high-resolution separation of complex plant extracts.

Ultra-Performance Liquid Chromatography (UPLC)

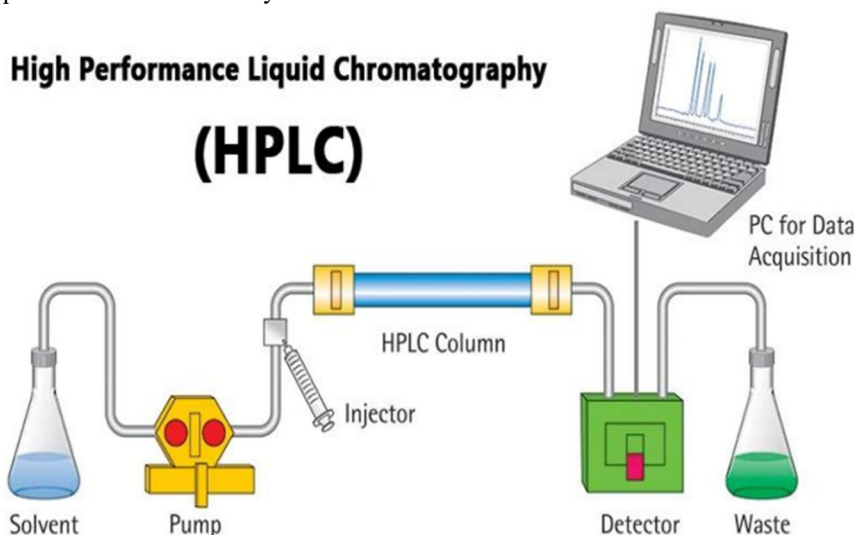
Ultra-Performance Liquid Chromatography (UPLC) is an advanced form of HPLC that provides significant improvements in chromatographic resolution, analysis speed, and sensitivity. UPLC uses very fine particles, typically smaller than 2.5 μm , which results in higher column efficiency and better resolution. Unlike conventional HPLC, efficiency in UPLC does not decrease at higher linear velocities or flow rates. As a result, UPLC allows faster separations with reduced solvent consumption and improved detection limits. These advantages make UPLC highly suitable for high-throughput pharmaceutical analysis and complex sample matrices.

Principle of High-Performance Liquid Chromatography (HPLC)

High-Performance Liquid Chromatography (HPLC) is an analytical separation technique in which purification and separation occur within a column containing a stationary phase and a mobile phase. The stationary phase consists of finely divided, porous particles packed inside the column, while the mobile phase is a liquid solvent or solvent mixture pumped through the column under high pressure. The sample is introduced into the flowing mobile phase through an



injector, usually via a sample loop. As the sample travels through the column, its components interact differently with the stationary phase and therefore migrate at different rates. This differential retention results in separation of the components. After elution from the column, the separated compounds are detected by a suitable detector and converted into electrical signals. These signals are processed by HPLC software to produce a chromatogram, which enables both identification and quantification of the analytes.



Types of HPLC and Columns Used in Analysis

HPLC columns play a central role in separation, as they determine efficiency, selectivity, and resolution. Most columns are packed with silica gel due to its small particle size, porosity, and chemical inertness. The material packed inside the column is known as the stationary phase. Based on the nature of the stationary and mobile phases, HPLC is classified into several types.

Normal Phase HPLC

Normal phase HPLC uses a polar stationary phase, typically silica, and a non-polar mobile phase such as hexane, chloroform, or methylene chloride. Water is not used due to its high polarity. Separation occurs based on polarity, where more polar compounds interact strongly with the stationary phase and elute later than less polar compounds. This technique is widely applied in pharmaceutical analysis.

Reverse Phase HPLC

Reverse phase HPLC is the most commonly used mode of HPLC. It employs a non-polar stationary phase, such as C8 or C18 bonded silica, and a polar mobile phase consisting of water mixed with methanol or acetonitrile. Separation is based on hydrophobic interactions, with non-polar compounds being retained longer. This method offers excellent reproducibility and is suitable for a wide range of analytes.

Ion Exchange HPLC

Ion exchange chromatography is used for the analysis of ionizable compounds. The stationary phase carries charged functional groups, while the mobile phase is a polar aqueous solution containing salts. Separation occurs due to electrostatic interactions between the analyte ions and the charged stationary phase. This method is particularly useful for inorganic ions, amino acids, and proteins.

Size Exclusion HPLC

Size exclusion chromatography separates molecules based on their molecular size. The stationary phase contains porous particles, allowing smaller molecules to enter the pores and elute later, while larger molecules elute earlier. Polymers and silica-based materials are commonly used. This technique is less frequently applied in pharmaceutical analysis but is useful for biomolecules and polymers.



Instrumentation of HPLC

An HPLC system consists of several essential components that work together to achieve effective separation and detection.

The **pump** delivers the mobile phase from the solvent reservoir to the column at high pressure while maintaining a constant and reproducible flow rate.

The **injector** introduces the sample into the mobile phase stream, commonly using a sampling loop or an autosampler for repetitive and precise injections.

The **column** is the site of separation and is typically housed in stainless steel to withstand high pressure and a wide range of solvents.

The **detector** monitors the eluent exiting the column and converts changes in composition into electronic signals. Common detectors include UV-Visible, fluorescence, and mass spectrometry detectors.

The **recorder or data system** processes detector signals and generates chromatograms, enabling peak integration, quantification, and data analysis.

The **degasser** removes dissolved gases from the mobile phase to prevent baseline noise and signal instability.

The **column heater (oven)** maintains a constant column temperature, improving reproducibility and resolution, especially for temperature-sensitive analyses.

Advantages of HPLC

HPLC offers high speed, efficiency, accuracy, and versatility. It provides exceptional precision in identifying and quantifying chemical components, making it a preferred analytical technique in both research and industry.

Limitations of HPLC

Despite its advantages, HPLC has certain limitations. The technique can be expensive due to the cost of instruments, columns, and organic solvents. Method development and operation require skilled personnel. Some compounds exhibit low sensitivity or irreversible adsorption, and volatile substances are better analyzed using gas chromatography.

Applications of HPLC

HPLC is widely applied in drug discovery, clinical analysis, proteomics, forensic chemistry, drug metabolism studies, environmental chemistry, diagnostic studies, cosmetic analysis, structural determination, and pharmaceutical analysis. It is also used for the identification of bile acid metabolites and therapeutic drug monitoring.

Pharmaceutical Applications

HPLC is used for tablet dissolution studies, shelf-life determination, identification of active ingredients, and pharmaceutical quality control.

Environmental Applications

It is applied for detection of phenolic compounds in drinking water, identification of pollutants in sediments, and biomonitoring of environmental contaminants.

Forensic Applications

HPLC assists in quantification of drugs in biological samples, identification of anabolic steroids, forensic analysis of textile dyes, and detection of cocaine and its metabolites.

Clinical Applications

Clinical uses include biochemical genetics, qualitative and quantitative analysis, therapeutic drug monitoring, analysis of antibiotics in blood plasma, estimation of bilirubin in hepatic disorders, and detection of neuropeptides.

Food and Flavor Analysis

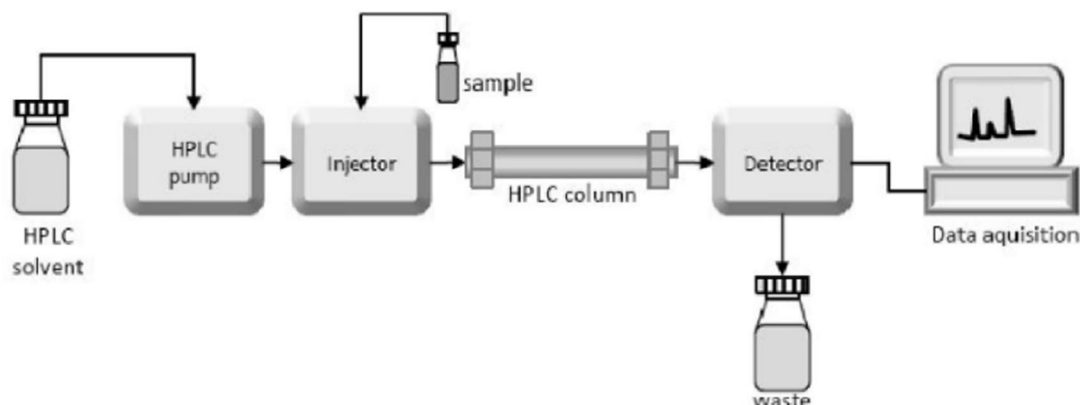
HPLC ensures quality control of soft drinks and water, analysis of beer, sugar determination in fruit juices, detection of polycyclic compounds in vegetables, and trace analysis of explosives in agricultural crops.

Construction of High-Performance Liquid Chromatography (HPLC)

Solvent Reservoir

Holds the mobile phase (single solvent or mixture). Solvents are usually filtered and degassed before use.





Degasser

Removes dissolved gases from the mobile phase to prevent bubble formation and baseline noise.

Pump

Delivers the mobile phase at high pressure with a constant and reproducible flow rate through the system.

Injector (Manual or Auto-Injector)

Introduces a precise volume of sample into the flowing mobile phase.

Chromatographic Column

Packed with stationary phase (commonly silica-based particles such as C18). Actual separation of sample components occurs here.

Column Oven (Heater)

Maintains a constant temperature to improve reproducibility and resolution.

Detector

Detects separated analytes as they elute from the column and converts them into electrical signals (e.g., UV-Vis, Fluorescence, MS).

Data System (Recorder/Computer)

Processes detector signals and generates a chromatogram for identification and quantification.

Working of High-Performance Liquid Chromatography (HPLC)

The working of HPLC is based on the differential interaction of sample components between a stationary phase and a mobile phase under high pressure.

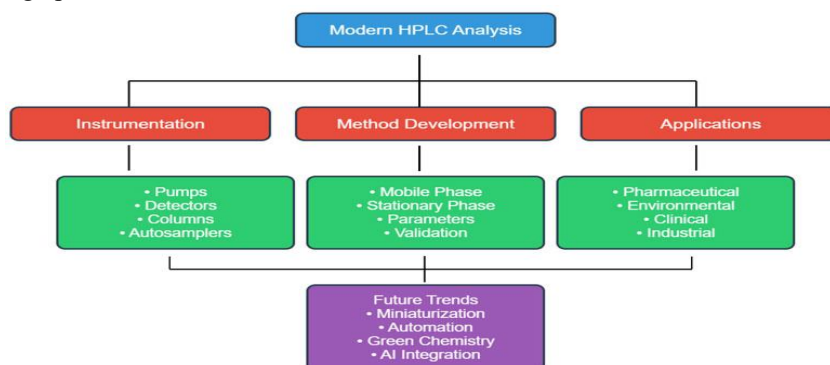


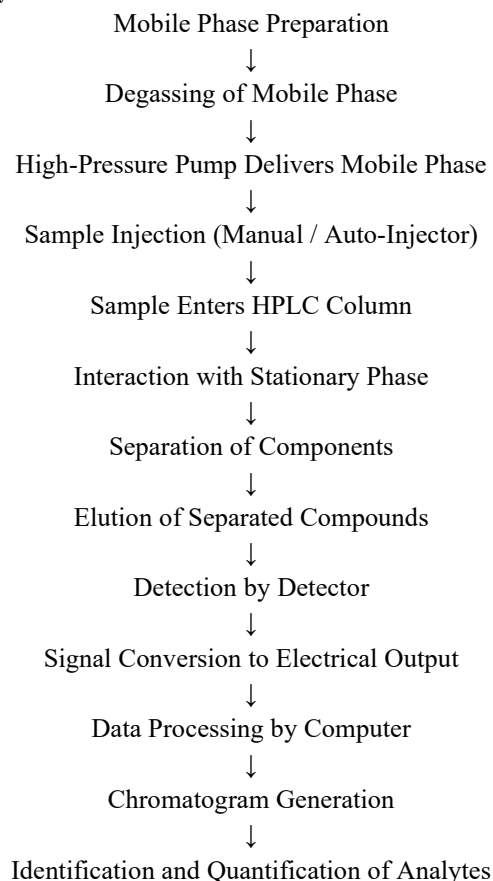
Figure 1: HPLC components and future trends

The mobile phase is pumped from the solvent reservoir through the system at high pressure. The sample is injected into the flowing mobile phase and carried into the column. Inside the column, different components of the sample interact differently with the stationary phase depending on their chemical properties. Components with stronger interactions are retained longer, while others elute faster. As the separated components exit the column, they pass through the detector,



which records their presence as peaks in the chromatogram. The retention time and peak area are used for identification and quantification of analytes.

Working Parameter / Methodology



II. RESULTS & DISCUSSION

This review highlights the continuous evolution of High-Performance Liquid Chromatography (HPLC) as a core analytical technique across pharmaceutical, environmental, clinical, and food analysis. The collected data demonstrate that conventional HPLC provides reliable separation, identification, and quantification of compounds with high accuracy and precision. However, recent technological developments have significantly enhanced chromatographic performance. Advanced HPLC techniques such as Rapid Resolution Liquid Chromatography (RRLC) and Ultra-Performance Liquid Chromatography (UPLC) show marked improvements in analysis speed, separation efficiency, resolution, and sensitivity compared to conventional HPLC. The use of sub-2 μm particle columns, higher operating pressures, optimized flow rates, and controlled column temperatures has resulted in reduced analysis time and lower solvent consumption without compromising resolution. Instrumentation advancements, including improved pumps, autosamplers, detectors, degassers, and data systems, have increased reproducibility and robustness. Hyphenated techniques such as HPLC-MS/MS have further enhanced detection limits and selectivity, enabling trace-level analysis in complex matrices. Overall, modern HPLC systems demonstrate superior performance in terms of throughput, reliability, and analytical capability.

The findings of this review clearly indicate that HPLC has transitioned from a conventional separation technique into a highly sophisticated analytical platform. Method development and validation remain critical for achieving accurate and reproducible results, particularly in regulated industries such as pharmaceuticals. Parameters such as mobile phase composition, stationary phase selection, temperature control, and detection techniques significantly influence chromatographic outcomes. Newer techniques such as RRLC and UPLC address the limitations of traditional HPLC by



improving speed and efficiency while maintaining high resolution. The adoption of smaller particle size columns and higher operating pressures has enabled faster separations and improved sensitivity. These developments are especially valuable in pharmaceutical quality control, impurity profiling, and pharmacokinetic studies where high throughput and reliability are required.

The integration of advanced detectors, particularly mass spectrometry, has expanded the applicability of HPLC to complex biological and environmental samples. Furthermore, trends toward automation, miniaturization, and green analytical chemistry have reduced solvent consumption and environmental impact. Despite these advantages, challenges such as high instrument cost and technical complexity remain, emphasizing the need for skilled operation and careful method optimization.

III. CONCLUSION

High-Performance Liquid Chromatography (HPLC) continues to be one of the most powerful and versatile analytical techniques in modern science. This review demonstrates that ongoing advancements in column technology, instrumentation, and detection systems have significantly enhanced chromatographic performance. Emerging techniques such as RRLC and UPLC provide faster analysis, higher resolution, improved sensitivity, and reduced solvent usage compared to conventional HPLC. HPLC plays a crucial role in pharmaceutical analysis, clinical diagnostics, environmental monitoring, forensic investigations, and food quality control. Rigorous method development and validation ensure compliance with regulatory standards and guarantee analytical reliability. With continued innovation in automation, hyphenated techniques, and sustainable practices, HPLC is expected to remain a cornerstone of analytical chemistry and continue to meet future analytical challenges effectively.

REFERENCES

- [1]. Kamble S, Agrawal S, Pagade S, Patil R, Chaugule N, Patil A. A review on high-performance liquid chromatography (HPLC). *Asian J Pharm Anal.* 2023;13(1):61–65.
- [2]. Rogatsky E. Modern high-performance liquid chromatography and HPLC 2016 international symposium. *J Chromatogr Sep Tech.* 2016;7:e135.
- [3]. Nardulli P, et al. A combined HPLC and LC-MS approach for evaluating drug stability in elastomeric devices: a challenge for sustainability in pharmacoconomics. *J Pharmacovigilance.* 2014;2:157.
- [4]. Harmita, et al. Optimization and validation of analytical method of cotrimoxazole in tablet and plasma *in vitro* by high-performance liquid chromatography. *J Bioanal Biomed.* 2012;4:26–29.
- [5]. Wiklund AE, Dag B, Brita S. Toxicity evaluation using intact sediments and sediment extracts. *Mar Pollut Bull.* 2005;50(6):660–667.
- [6]. Kwok YC, Hsieh DPH, Wong PK. Toxicity identification evaluation of pore water from contaminated marine sediments collected from Hong Kong waters. *Mar Pollut Bull.* 2005;51(8–12):1085–1091.
- [7]. Yu H, Jing C, Cui Y, Shang H, Ding Z, Jin H. Application of toxicity identification evaluation procedures on wastewaters and sludge from a municipal sewage treatment works with industrial inputs. *Ecotoxicol Environ Saf.* 2004;57(3):426–430.
- [8]. Bounine JP, Tardif B, Beltran P, Mazzo DJ. High-performance liquid chromatographic stability-indicating determination of zopiclone in tablets. *J Chromatogr.* 1994;677(1):87–93.
- [9]. Abidi SL. High-performance liquid chromatography of phosphatidic acids and related polar lipids. *J Chromatogr.* 1991;587:193–203.
- [10]. Stubbs C, Kanfer I. Stability-indicating high-performance liquid chromatographic assay of erythromycin estolate in pharmaceutical dosage forms. *Int J Pharm.* 1990;3(2):113–119.
- [11]. Shah AJ, Adlard MW, Stride JD. A sensitive assay for clavulanic acid and sulbactam in biological fluids by high-performance liquid chromatography with precolumn derivatization. *J Pharm Biomed Anal.* 1990;5:437–443.
- [12]. Rodenas V, Garcia MS, Sanchez-Pedreno C, Albero MI. Flow-injection spectrophotometric determination of frusemide or sulphathiazole in pharmaceuticals. *J Pharm Biomed Anal.* 1997;15:1687–1693.



- [13]. Bergh JJ, Breytenbach JC. Stability-indicating high-performance liquid chromatographic analysis of trimethoprim in pharmaceuticals. *J Chromatogr.* 1987;387:528–531.
- [14]. Bowden RE, Madsen PO. High-pressure liquid chromatographic assay of sulbactam in plasma, urine, and tissue. *Antimicrob Agents Chemother.* 1986;30:231–233.
- [15]. MacNeil L, Rice JJ, Muhammad N, Lauback RG. Stability-indicating liquid chromatographic determination of cefapirin, desacetylcefapirin, and cefapirin lactone in sodium cefapirin bulk and injectable formulations. *J Chromatogr.* 1986;361:285–290.
- [16]. Fredj G, Paillet-Aussel MF, Brouard A, Barreteau H, Divine C, Micaud M. Determination of sulbactam in biological fluids by high-performance liquid chromatography. *J Chromatogr.* 1986;383:218–222.
- [17]. Haginaka J, Yasuda H, Uno T, Nakagawa T. Alkaline degradation and determination by high-performance liquid chromatography. *Chem Pharm Bull.* 1984;32:2752–2758.
- [18]. Lauback RG, Rice JJ, Bleiberg B, Muhammad N, Hanna SA. Specific high-performance liquid chromatographic determination of ampicillin in bulk, injectables, capsules, and oral suspensions by reversed-phase ion-pair chromatography. *J Liq Chromatogr.* 1984;7(6):1243–1265.
- [19]. Ayerton J. Assay of ceftazidime in biological fluids using high-pressure liquid chromatography. *J Antimicrob Chemother.* 1981;8:227–231.
- [20]. Tweeten TN, Euston CB. Advances in food technology. *Food Technol.* 1980;December:29–35.
- [21]. Christie WW, Gill S, Nordbäck J, Itabashi Y, Sanda S, Slabas AR. New procedures for rapid screening of leaf lipid components from *Arabidopsis*. *Phytochem Anal.* 1998;9:53–57.
- [22]. Belanger JM, Paré JJ, Sigouin M. High-performance liquid chromatography (HPLC): principles and applications. In: *Techniques and Instrumentation in Analytical Chemistry*. Vol. 18. Elsevier; 1997:37–59.
- [23]. Gorhe SG, Pawar GR. A review on high-performance liquid chromatography (HPLC). *Int J Adv Sci Res Eng Trends.* 2018;3(1):1–6.
- [24]. Saxby MJ. Developments in food analysis techniques. In: King RD, ed. *Developments in Food Analysis Techniques*. Vol. 1. London: Applied Science Publishers; 1978:125.

