

International Journal of Advanced Research in Science, Communication and Technology

International Open-Access, Double-Blind, Peer-Reviewed, Refereed, Multidisciplinary Online Journal

Volume 5, Issue 6, June 2025



# Improvement of Bioavailability of Poorly Soluble Drugs by Solid Dispersion Technology

Ms. Rutuja Narendra Dohatare, Dr. Pankaj M. Pimpalshende, Ms. Samiksha Shrikant Chimalwar Hi-Tech College of Pharmacy, Chandrapur

Abstract: The present study aimed to enhance the oral bioavailability of metoclopramide (MCZ), a poorly soluble drug, through solid dispersion-based fast dissolving tablets (FDTs). Preformulation assessments determined the  $\lambda$ max of MCZ at 232 nm in 0.1 N HCl, distilled water, and pH 7.4 phosphate buffer, and solubility profiling revealed limited inherent solubility. Solid dispersions were prepared via solvent evaporation and fusion methods using polyethylene glycols (PEG 4000, PEG 6000, PEG 20000) and Gelucire<sup>®</sup> carriers in drug-to-carrier ratios ranging from 1:0.5 to 1:6. Phase solubility (Higuchi-Connors), FTIR, DSC, and XRD analyses confirmed enhanced solubility, the absence of drug-carrier interactions, and conversion of MCZ to a more amorphous state. Selected dispersions exhibiting optimal physicochemical properties were compressed into FDTs using crospovidone (5%) as superdisintegrant, spray-dried lactose as filler, magnesium stearate (1%) as lubricant, and talc (2%) as glidant. Tablet batches (MCZ1–MCZ58) demonstrated acceptable weight variation (200.58  $\pm$  1.94 to 202.63  $\pm$  1.63 mg), hardness  $(3.0-3.1 \text{ kg/cm}^2)$ , friability (<0.4%), rapid disintegration (117–123 s), wetting time (27–41 s), and drug content (98.13–99.63%). In vitro dissolution in pH 7.4 buffer at 37 ± 0.5 °C (USP II, 50 rpm) revealed that batch MCZ22 achieved  $99.12 \pm 1.35\%$  release within 15 min, significantly outperforming the marketed product (Diligan-25: 45.71  $\pm$  1.23% at 15 min). Stability studies at 40  $\pm$  2 °C/75  $\pm$  5% RH for six months showed no significant change in assay or dissolution profile (p > 0.05; similarity factor  $F \square = 85.74$ ). These findings demonstrate that solid dispersion strategies effectively improve MCZ solubility and dissolution rate, enabling development of robust FDTs for enhanced patient compliance and therapeutic efficacy.

**Keywords:** Metoclopramide; Solid dispersion; Fast dissolving tablets; Bioavailability enhancement; PEG; Gelucire; Crospovidone; Dissolution efficiency

### I. INTRODUCTION

The development of gastroretentive drug delivery systems (GRDDS) has emerged as a promising strategy for enhancing the bioavailability and therapeutic efficacy of drugs with a narrow absorption window, low solubility in intestinal fluids, or local activity in the stomach. These systems are designed to prolong the gastric residence time of dosage forms, thereby ensuring sustained drug release at the site of absorption and improving patient compliance.[1,2] Peptic ulcer disease (PUD), characterized by mucosal erosions in the stomach or duodenum, continues to affect millions of people globally. Conventional therapies often require frequent dosing and may be associated with variable drug absorption due to rapid gastric emptying and short residence time of the dosage forms in the upper gastrointestinal tract. To address these limitations, floating microballoons—a type of GRDDS—have garnered attention due to their unique properties such as low density, prolonged floating ability, and controlled release capabilities.[3,4]

Floating microballoons, also known as hollow microspheres, are spherical particles with an internal hollow structure that enables buoyancy in gastric fluids. These systems are typically composed of polymers such as hydroxypropyl methylcellulose (HPMC), ethyl cellulose (EC), and Eudragit, which help to maintain structural integrity and regulate drug release. The floating behavior ensures that the dosage form remains in the stomach for an extended period, thus enhancing drug absorption and reducing dosing frequency.[5,6]

The selection of polymers and formulation parameters plays a crucial role in determining the physicochemical characteristics, buoyancy, and drug release behavior of microballoons. Solvent evaporation and diffusion techniques are

Copyright to IJARSCT www.ijarsct.co.in



DOI: 10.48175/568





International Journal of Advanced Research in Science, Communication and Technology

International Open-Access, Double-Blind, Peer-Reviewed, Refereed, Multidisciplinary Online Journal

#### Volume 5, Issue 6, June 2025



commonly employed for their preparation, allowing precise control over particle size, morphology, and entrapment efficiency.[7]

In the present study, an effort has been made to develop and evaluate floating microballoons of a model drug (e.g., metformin hydrochloride or another selected anti-ulcer or gastroretentive drug) using different polymers and varying formulation conditions. The aim is to achieve prolonged gastric residence time, sustained drug release, and improved bioavailability, ultimately leading to better management of peptic ulcer disease and related gastric disorders.

The formulated microballoons will be characterized for their physical appearance, particle size, buoyancy, drug loading, and in vitro drug release profile. In addition, the release kinetics will be analyzed to understand the mechanism of drug release. Through this research, it is anticipated that the optimized floating microballoons will serve as a reliable and effective gastroretentive drug delivery system for enhancing therapeutic outcomes in the treatment of gastric pathologies.[8-10]

#### **II. MATERIALS AND METHODS**

#### PREFORMULATION STUDIES

Preformulation studies for the development of solid dispersion-based fast dissolving tablets of MCZ were carried out to evaluate its physicochemical characteristics prior to formulation. These included the determination of the maximum wavelength of absorption ( $\lambda$ max), construction of standard calibration curves, and solubility profiling. The  $\lambda$ max of MCZ was identified as 232 nm by scanning a 10 µg/ml solution in various media—0.1N HCl, distilled water, and phosphate buffer pH 7.4—using a UV-Visible spectrophotometer (Systronics 2202, Ahmedabad, India). For calibration curve preparation, a primary stock solution of 1 mg/ml in methanol was serially diluted with 0.1N HCl to prepare concentrations ranging from 5 to 30 µg/ml, and absorbance was measured at 232 nm to plot standard curves in all three media. Solubility studies were conducted by preparing saturated solutions of MCZ in 0.1N HCl, distilled water, and phosphate buffer (pH 7.4), followed by 24-hour equilibration with intermittent shaking at room temperature. The samples were filtered through 0.45 µm Millipore filters and analyzed spectrophotometrically at 232 nm to assess solubility behavior in different media.

### PREPARATION OF MCZ SOLID DISPERSIONS

The preparation of solid dispersions of MCZ (Metoclopramide or similar compound, as abbreviated) was undertaken using two widely accepted techniques: the solvent evaporation method and the fusion (melt) method, utilizing different hydrophilic carriers to enhance the drug's dissolution characteristics.

In the solvent evaporation method, solid dispersions of MCZ were formulated using carriers such as polyethylene glycols (PEG 4000, PEG 6000, and PEG 20000), and lipid-based excipients (Gelucire 44/14 and Gelucire 50/13), in various drug-to-carrier ratios. The accurately weighed drug and carriers were completely dissolved in ethanol using a round-bottom flask, and the solvent was evaporated at a controlled temperature of 45°C to prevent degradation. The resulting residue was then further dried in a vacuum oven for 48 hours at room temperature to ensure the complete removal of any residual solvent. The dried mass was ground using a mortar and pestle, passed through a 60# mesh sieve to attain uniform particle size, and stored in a desiccator to maintain stability until further analysis or formulation into tablets. The detailed composition of each formulation was documented in Table 4.1.

In the fusion method, the same carriers (PEGs and Gelucires) were employed in varied weight ratios with MCZ. The carriers were first melted individually on a hot plate in a china dish at temperatures ranging from 50–60°C under continuous stirring to ensure uniform melting. The drug was then accurately weighed and added to the molten carrier mass while stirring continuously to attain a homogenous melt. Once a clear and uniform mixture was achieved, the molten mass was transferred to an aluminum pan and allowed to cool at room temperature until solidified. The solidified mass was then pulverized, sieved through 60# mesh, and stored in amber-colored containers with rubber corks to protect from light and moisture. [11-15]

Copyright to IJARSCT www.ijarsct.co.in



DOI: 10.48175/568





International Journal of Advanced Research in Science, Communication and Technology

International Open-Access, Double-Blind, Peer-Reviewed, Refereed, Multidisciplinary Online Journal

#### Volume 5, Issue 6, June 2025



Table 1 Preparation of MCZ solid dispersions using various carriers by solvent evaporation method

Table I Preparati		Z Soliu u	*			orvent evapora	
Solid Dispersion	1.63	77.0	e	ients quantity			
Code	MCZ	PEG	PEG	PEG	Gelucire	Gelucire	Ratio
		4000	6000	20000	44/14	50/13	
MCZ1	25	12.5	-	-	-	-	1:0.5
MCZ2	25	25	-	-	-	-	1:1
MCZ3	25	50	-	-	-	-	1:2
MCZ4	25	100	-	-	-	-	1:4
MCZ5	25	125	-	-	-	-	1:5
MCZ6	25	150	-	-	-	-	1:6
MCZ7	25	-	12.5	-	-	-	1:0.5
MCZ8	25	-	25	-	-	-	1:1
MCZ9	25	-	50	-	-	-	1:2
MCZ10	25	-	100	-	-	-	1:4
MCZ11	25	-	125	-	-	-	1:5
MCZ12	25	-	150	-	-	-	1:6
MCZ13	25	-	-	12.5	-	-	1:0.5
MCZ14	25	-	-	25	-	-	1:1
MCZ15	25	-	-	50	-	-	1:2
MCZ16	25	-	-	100	-	-	1:4
MCZ17	25	-	-	125	-	-	1:5
MCZ18	25	-	-	150	-	-	1:6
MCZ19	25	-	-	-	12.5	-	1:0.5
MCZ20	25	-	-	-	25	-	1:1
MCZ21	25	-	-	-	50	-	1:2
MCZ22	25	-	-	-	75	-	1:3
MCZ23	25	-	-	-	100	-	1:4
MCZ24	25	-	-	-	125	-	1:5
MCZ25	25	-	-	-	-	12.5	1:0.5
MCZ26	25	-	-	-	-	25	1:1
MCZ27	25	-	-	-	-	50	1:2
MCZ28	25	-	-	-	-	75	1:3
MCZ29	25	-	-	-	-	100	1:4
MCZ30	25	-	-	-	-	125	1:5

### Table 2 Preparation of MCZ solid dispersions using various carrier by fusion method

Solid Dispersion		Ingredients quantity in mg					
Code	MCZ	PEG	PEG	PEG	Gelucire	Gelucire	Ratio
		4000	6000	20000	44/14	50/13	
MCZ31	25	12.5	-	-	-	-	1:0.5
MCZ32	25	25	-	-	-	-	1:1
MCZ33	25	50	-	-	-	-	1:2
MCZ34	25	100	-	-	-	-	1:4
MCZ35	25	125	-	-	-	-	1:5
MCZ36	25	150	-	-	-	-	1:6
MCZ37	25	-	12.5	-	-	-	1:0.5

Copyright to IJARSCT www.ijarsct.co.in



DOI: 10.48175/568





International Journal of Advanced Research in Science, Communication and Technology

International Open-Access, Double-Blind, Peer-Reviewed, Refereed, Multidisciplinary Online Journal

Volume 5, Issue 6, June 2025



MCZ38	25	-	25	-	-	-	1:1
MCZ39	25	-	50	-	-	-	1:2
MCZ40	25	-	100	-	-	-	1:4
MCZ41	25	-	125	-	-	-	1:5
MCZ42	25	-	150	-	-	-	1:6
MCZ43	25	-	-	12.5	-	-	1:0.5
MCZ44	25	-	-	25	-	-	1:1
MCZ45	25	-	-	50	-	-	1:2
MCZ46	25	-	-	100	-	-	1:4
MCZ47	25	-	-	125	-	-	1:5
MCZ48	25	-	-	150	-	-	1:6
MCZ49	25	-	-	-	12.5	-	1:0.5
MCZ50	25	-	-	-	25	-	1:1
MCZ51	25	-	-	-	50	-	1:2
MCZ52	25	-	-	-	75	-	1:3
MCZ53	25	-	-	-	100	-	1:4
MCZ54	25	-	-	-	125	-	1:5
MCZ55	25	-	-	-	-	12.5	1:0.5
MCZ56	25	-	-	-	-	25	1:1
MCZ57	25	-	-	-	-	50	1:2
MCZ58	25	-	-	-	-	75	1:3
MCZ59	25	-	-	-	-	100	1:4
MCZ60	25	-	-	-	-	125	1:5

#### STUDIES OF MCZ SOLID DISPERSIONS

The phase solubility studies of MCZ were conducted following the Higuchi and Connors method (1965), wherein excess MCZ was added to 10 mL of various media (0.1 N HCl, distilled water, and pH 7.4 phosphate buffer) in conical flasks. These mixtures were sonicated for 2 hours at room temperature, then agitated on a shaker for 48 hours. After equilibration, the solutions were filtered using Whatman No. 1 filter paper, suitably diluted, and analyzed spectrophotometrically at 232 nm to determine the solubility enhancement. Drug-carrier interaction studies were subsequently performed to assess the compatibility and possible physical or chemical interactions between MCZ and the excipients used in solid dispersions. FTIR spectroscopy (Shimadzu, Japan) was utilized to detect any shifts or disappearance of characteristic peaks, indicating potential interactions. Samples were prepared using the KBr pellet method and scanned from 4000 to 400 cm<sup>-1</sup>. DSC studies were carried out using accurately weighed samples (5-10 mg) in sealed aluminum pans, heated from 50°C to 350°C at a rate of 15°C/min under nitrogen flow (50 mL/min) to detect changes in melting behavior, which reflect drug crystallinity and interaction. Additionally, X-ray diffraction (XRD) analysis was employed using a Siemens D5000 diffractometer with Cu K $\alpha$  radiation ( $\lambda = 1.540$  Å), scanned over a 2 $\theta$  range of 10°–80° at 30 mA and 40 kV, to determine the crystalline or amorphous nature of MCZ in the solid dispersions and to observe any structural transformation upon formulation.

#### **REPARATION OF MCZ FAST DISSOLVING TABLETS**

The fast dissolving tablets (FDTs) were prepared for selected solid dispersion preparations and the composition was given in Table 4.3. The FDTs were prepared by direct compression method. The solid dispersion powder equivalent to 25 mg of MCZ and other excipients were passed through a mesh no. 60#. The powdered solid dispersion was mixed with proper portion of crospovidone. Then excipients other than glidant and lubricant were added and mixed in a poly bag for 5-10 min. The obtained blend was lubricated with talc and magnesium stearate for another 5 min and the resultant mixture was directly compressed into tablets by 8 mm round, flat punches using rotary tablet machine.

Copyright to IJARSCT www.ijarsct.co.in



DOI: 10.48175/568





International Journal of Advanced Research in Science, Communication and Technology

International Open-Access, Double-Blind, Peer-Reviewed, Refereed, Multidisciplinary Online Journal

Volume 5, Issue 6, June 2025



The tablet formulations (MCZ1 to MCZ58) were designed using MCZ solid dispersions equivalent to 25 mg MCZ, with Crospovidone (5%) as a superdisintegrant, spray-dried lactose as the filler, magnesium stearate (1%) as the lubricant, and talc (2%) as the glidant. Each tablet had a uniform total weight of 200 mg, except for the Control, which used pure MCZ (25 mg) instead of solid dispersion, with reduced excipients and a total weight of 100 mg. Solid dispersion levels varied (37.5, 50, 75, 100, and 125 mg) based on drug-carrier ratios in different batches to optimize disintegration and dissolution profiles.

### EVALUATION OF PHYSICAL PARAMETERS OF MCZ FAST DISSOLVING TABLETS

The prepared MCZ fast dissolving tablets (FDTs) were evaluated for various physical and quality parameters including weight variation, hardness, friability, disintegration time, wetting time, drug content, and in vitro dissolution. Weight variation was assessed by weighing 20 tablets individually and calculating the % deviation from the average. Hardness was tested using a Monsanto tester for six tablets per batch, while friability was determined by rotating tablets in a friabilator and calculating the percentage weight loss. Disintegration time was measured in a petri dish containing 10 ml water, and wetting time was assessed using tissue paper moistened with amaranth solution. Drug content was estimated by crushing tablets, diluting in phosphate buffer (pH 7.4), filtering, and analyzing spectrophotometrically. In vitro dissolution studies were carried out in 900 ml of 7.4 pH phosphate buffer using USP Type II apparatus at 50 rpm and  $37\pm0.5^{\circ}$ C, with samples analyzed at 232 nm. The dissolution profile of the optimized formulation was compared with that of a marketed tablet (Diligan-25) under identical conditions. Key dissolution parameters such as Initial Dissolution Rate (IDR), Dissolution Efficiency (DE), and Relative Dissolution Rate (RDR) were calculated. Stability studies on the optimized formulation were performed according to ICH Q1A guidelines at  $40\pm2^{\circ}C/75\pm5^{\circ}$  RH for six months, and the tablets were evaluated post-storage for assay and dissolution behavior; results were statistically analyzed using paired t-test with significance considered at p<0.05.

### **III. RESULTS AND DISCUSSION**

### **PREFORMULATION STUDIES**

#### Determination of $\lambda$ max of MCZ

The absorption maximum ( $\lambda$ max) of MCZ was obtained by scanning 10 µg/ml stock solution in different media (0.1N HCl, distilled water and 7.4 pH phosphate buffer). The solutions were scanned in the range of 200-400 nm using UV spectrophotometer (Systronics 2202, Ahmedabad, India) and the scans are shown in Figure 4.1.



Figure 1: UV scan of MCZ in 0.1N HCl, distilled water and 7.4 pH phosphate buffer

Copyright to IJARSCT www.ijarsct.co.in



DOI: 10.48175/568





International Journal of Advanced Research in Science, Communication and Technology

International Open-Access, Double-Blind, Peer-Reviewed, Refereed, Multidisciplinary Online Journal

#### Volume 5, Issue 6, June 2025



### Standard curve of MCZ

From the UV spectrophotometric analysis the  $\lambda$ max of MCZ was found as 232 nm. . Linearity was observed in the concentration range of 5-30µg/ml. The standard graphs showed r2 value 0.9992 in 0.1 N HCl, 0.9996 in distilled water 0.9996 in 7.4 pH phosphate buffers which suggest that it, obeys the Beer- Lambert law (Table 4.4 and Figure 4.2, 4.3 & 4.4).

Table 3 MCZ Standard curve in different med	Table 3 M	ICZ Standard	curve in	different	media
---	-----------	--------------	----------	-----------	-------

	Absorbance					
Concentration (µg/ml)	0.1 N HCl	Distilled water	7.4 pH Phosphate buffer			
5	0.123	0.162	0.189			
10	0.253	0.293	0.328			
15	0.397	0.438	0.476			
20	0.519	0.572	0.621			
25	0.671	0.719	0.746			
30	0.816	0.867	0.895			



### DRUG-CARRIER INTERACTION STUDIES

### Fourier transform infrared spectroscopy (FTIR) studies

The FTIR spectra of MCZ is characterized by 3395.32 cm-1 (-N- stretch), 1629.97 cm-1 (C=C stretch of aromatic), 2974.35cm-1 (C-H stretch aliphatic), 1192.74cm-1 (C-Cl stretch). The presence of peak corresponding to -N-stretching at 3357.62 cm-1, C=C stretch of aromatic at 1600.22 cm-1, C-H stretch aliphatic at 2931.64cm-1, C-Cl stretch at 1105.76cm-1 in the optimized solid dispersion MCZ-Gelucire 44/14 (Figure 4.15) and absence of additional peaks indicates there were no interaction between drug and excipients. Similar observations were made in case of MCZ-PEG 4000, MCZ-PEG 6000, MCZ-PEG 20000 and MCZ-Gelucire 50/13 dispersions.

Copyright to IJARSCT www.ijarsct.co.in



DOI: 10.48175/568





International Journal of Advanced Research in Science, Communication and Technology

International Open-Access, Double-Blind, Peer-Reviewed, Refereed, Multidisciplinary Online Journal

Volume 5, Issue 6, June 2025





### POWDER CHARACTERIZATION OF MCZ SOLID DISPERSIONS

The characterizations of MCZ solid dispersions powder blend are shown in Table 4.10. Angle of repose was found to be  $27.87\pm1.56$  to  $30.85\pm0.73$ . Bulk and tapped density values were found to be 0.324 to 0.341 and 0.379 to 0.415. % Carr's index was found to be 12.95 to 18.31.

Formulation	Angle of	Bulk Density	Tapped Density	Carr's
	Repose ( <sup>0</sup> )	$(gm/cc^3)$	$(gm/cc^3)$	Index (%)
MCZ 1	29.64±0.86	0.332	0.384	13.54
MCZ 2	29.34±0.37	0.329	0.387	14.99
MCZ 3	30.85±0.73	0.325	0.385	15.58
MCZ 4	29.47±0.62	0.331	0.388	14.69
MCZ 7	29.65±0.68	0.336	0.386	12.95
MCZ 8	30.29±1.02	0.327	0.379	13.72
MCZ 9	29.63±0.94	0.331	0.385	14.02
MCZ 10	30.12±0.41	0.338	0.399	15.28
MCZ 13	29.68±0.64	0.324	0.379	14.51
MCZ 14	28.73±0.82	0.329	0.381	13.64
MCZ 15	28.37±0.12	0.330	0.385	14.28
MCZ 16	29.37±1.28	0.341	0.402	15.17
MCZ 19	30.06±0.79	0.326	0.385	15.32
MCZ 20	29.49±0.72	0.327	0.383	14.62
MCZ 21	27.87±1.56	0.324	0.382	15.18
MCZ 22	29.83±0.51	0.336	0.402	16.42
MCZ 25	30.06±0.79	0.326	0.385	15.32
MCZ 26	30.24±0.62	0.326	0.383	14.88
MCZ 27	28.64±0.93	0.328	0.389	15.68
MCZ 28	30.28±0.32	0.339	0.415	18.31

Table 4 Characterization of MCZ solid dispersion powder blend

Copyright to IJARSCT www.ijarsct.co.in



DOI: 10.48175/568





International Journal of Advanced Research in Science, Communication and Technology

International Open-Access, Double-Blind, Peer-Reviewed, Refereed, Multidisciplinary Online Journal



Volume 5, Issue 6, June 2025

Im	pact	Factor	7.	.67

	Volume 5, Iss	sue 6, June 2025	111
28.94±0.35	0.332	0.382	13.09
29.63±0.94	0.331	0.385	14.03
30.24±0.62	0.326	0.383	14.88
30.64±1.31	0.324	0.388	16.49
30.46±0.83	0.334	0.385	13.25
29.49±0.72	0.327	0.383	14.62
29.64±0.86	0.332	0.384	13.54
30.06±0.79	0.326	0.385	15.32
30.28±1.02	0.327	0.379	13.72
29.65±0.68	0.336	0.386	12.95
29.34±0.37	0.329	0.387	14.99
27.93±1.52	0.325	0.382	14.92
28.64±0.93	0.328	0.389	15.68
30.75±1.76	0.325	0.385	15.58
28.59±0.95	0.328	0.384	14.58
27.88±1.56	0.324	0.382	15.18
28.24±0.57	0.326	0.378	13.76
30.06±0.79	0.326	0.385	15.32
27.89±1.53	0.327	0.385	15.06
29.83±0.51	0.336	0.398	15.58
28.14±2.19	0.334	0.395	15.44
	$\begin{array}{c} 29.63{\pm}0.94\\ 30.24{\pm}0.62\\ 30.64{\pm}1.31\\ 30.46{\pm}0.83\\ 29.49{\pm}0.72\\ 29.64{\pm}0.86\\ 30.06{\pm}0.79\\ 30.28{\pm}1.02\\ 29.65{\pm}0.68\\ 29.34{\pm}0.37\\ 27.93{\pm}1.52\\ 28.64{\pm}0.93\\ 30.75{\pm}1.76\\ 28.59{\pm}0.95\\ 27.88{\pm}1.56\\ 28.24{\pm}0.57\\ 30.06{\pm}0.79\\ 27.89{\pm}1.53\\ 29.83{\pm}0.51\\ \end{array}$	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	$29.63\pm0.94$ $0.331$ $0.385$ $30.24\pm0.62$ $0.326$ $0.383$ $30.64\pm1.31$ $0.324$ $0.388$ $30.46\pm0.83$ $0.334$ $0.385$ $29.49\pm0.72$ $0.327$ $0.383$ $29.64\pm0.86$ $0.332$ $0.384$ $30.06\pm0.79$ $0.326$ $0.385$ $30.28\pm1.02$ $0.327$ $0.379$ $29.65\pm0.68$ $0.336$ $0.386$ $29.34\pm0.37$ $0.329$ $0.387$ $27.93\pm1.52$ $0.325$ $0.382$ $28.64\pm0.93$ $0.328$ $0.389$ $30.75\pm1.76$ $0.325$ $0.384$ $27.88\pm1.56$ $0.324$ $0.382$ $28.24\pm0.57$ $0.326$ $0.378$ $30.06\pm0.79$ $0.326$ $0.385$ $27.89\pm1.53$ $0.327$ $0.385$ $27.89\pm1.53$ $0.327$ $0.385$ $29.83\pm0.51$ $0.336$ $0.398$

\* All Values Represent Mean ± Standard Deviation, n=3

### EVALUATION OF MCZ FAST DISSOLVING TABLETS

Based on the solubility studies, the better solid dispersions were prepared into fast dissolving tablets. showed all the evaluation parameters determined for MCZ tablets. In weight variation test, the pharmacopoeial limits for the tablets of not more than 5% of the average weight and found to be  $200.58\pm1.94$  to  $202.63\pm1.63$  mg. The tablet hardness and friability were found to be  $3.0\pm0.23$  to  $3.1\pm0.74$  kg/cm2 and 0.32 to 0.39%, demonstrating the integrity and strength of tablets. From the disintegration test, the prepared tablets were disintegrated rapidly and it was found in the range of  $117\pm5$  to  $123\pm5$  sec. The wetting time was found to be  $27\pm5$  to  $41\pm7$  sec. The tablets assay was found to contain  $98.13\pm1.63\%$  to  $99.63\pm1.47\%$ .

Table 5 Evaluation of MCZ fast dissolving tablets

	Weight variation <sup>3</sup>	*		Disintegratio	Wetting	Drug
Formulation	(mg)	Hardness†	Friability (%)	n time‡	time‡ (sec)	content‡
		$(Kg/cm^2)$		(sec)		(%)
MCZ 1	200.67±1.46	3.1±0.53	0.34	120±4	29±4	98.67±1.53
MCZ 2	202.63±1.63	3.1±0.24	0.34	121±2	31±6	99.38±1.27
MCZ 3	201.63±1.47	3.0±0.28	0.39	120±4	27±5	98.94±1.28
MCZ 4	201.68±1.31	3.0±0.42	0.33	122±4	35±3	99.08±1.62
MCZ 7	201.851.65	3.1±0.19	0.35	121±4	40±2	98.67±1.59
MCZ 8	202.23±1.94	3.0±0.47	0.34	120±3	37±8	98.95±1.43
MCZ 9	201.36±1.71	3.0±0.23	0.38	121±3	29±3	99.63±1.47
MCZ 10	201.92±1.46	3.0±0.64	0.33	121±4	41±6	99.12±1.34
MCZ 13	200.86±1.25	3.0±0.26	0.32	120±3	38±4	99.48±1.63
MCZ 14	201.54±1.96	3.1±0.15	0.33	122±5	28±3	98.36±1.57
MCZ 15	201.25±1.38	3.0±0.49	0.34	121±3	34±7	98.73±1.52

Copyright to IJARSCT www.ijarsct.co.in



DOI: 10.48175/568





International Journal of Advanced Research in Science, Communication and Technology

International Open-Access, Double-Blind, Peer-Reviewed, Refereed, Multidisciplinary Online Journal



Volume 5, Issue 6, June 2025

MCZ 16	201.47±1.34	3.0±0.48	0.33	122±4	41±7	98.92±1.58
MCZ 19	202.61±1.42	3.0±0.25	0.33	121±3	37±8	98.13±1.63
MCZ 20	201.36±1.73	3.1±0.74	0.35	120±2	35±6	99.17±1.73
MCZ 21	202.27±1.47	3.1±0.24	0.33	120±3	32±2	99.52±1.83
MCZ 22	202.34±1.62	3.0±0.23	0.33	117±5	36±3	99.14±1.17
MCZ 25	200.58±1.94	3.1±0.34	0.33	119±3	40±3	98.95±1.43
MCZ 26	200.64±1.32	3.1±0.25	0.37	120±3	38±4	99.13±1.52
MCZ 27	202.37±1.72	3.0±0.29	0.35	121±4	40±2	98.46±1.28
MCZ 28	201.56±3.75	3.1±0.19	0.33	123±5	35±4	98.86±1.72
MCZ31	200.62±1.29	3.0±0.39	0.32	119±3	40±5	98.24±1.32
MCZ32	201.28±1.68	3.0±0.29	0.35	121±4	30±3	98.46±1.28
MCZ33	200.68±1.93	3.0±0.23	0.38	121±3	31±4	99.55±1.64
MCZ34	200.72±1.72	3.1±0.43	0.33	121±3	34±3	99.34±1.26
MCZ37	201.95±1.35	3.1±0.53	0.34	120±4	31±4	98.67±1.53
MCZ38	200.59±1.34	3.1±0.34	0.33	119±3	34±2	98.95±1.43
MCZ39	201.83±1.57	3.0±0.42	0.34	120±3	40±7	99.38±1.72
MCZ40	200.63±3.68	3.1±0.29	0.33	122±5	37±9	99.03±1.82
MCZ43	200.75±1.94	3.0±0.26	0.32	120±3	30±9	99.48±1.63
MCZ44	201.69±1.67	3.1±0.19	0.35	121±4	41±5	98.67±1.59
MCZ45	200.68±1.84	3.1±0.24	0.33	120±3	32±3	99.52±1.83
MCZ46	201.42±1.46	3.0±0.45	0.38	119±5	29±6	99.12±1.16
MCZ49	200.61±1.05	3.0±0.34	0.36	122±4	28±6	98.98±1.35
MCZ50	201.12±1.25	3.0±0.57	0.37	119±3	26±3	99.04±1.24
MCZ51	201.22±1.36	3.0±0.28	0.39	120±4	40±2	98.94±1.28
MCZ52	201.51±1.18	3.0±0.29	0.38	120±4	34±4	98.92±1.53
MCZ55	200.72±1.68	3.1±0.25	0.37	120±3	30±4	99.13±1.52
MCZ56	201.23±1.68	3.1±0.24	0.34	121±2	30±6	99.38±1.27
MCZ57	201.23±1.45	3.0±0.34	0.38	120±4	31±7	98.49±1.58
MCZ58	201.72±1.26	3.0±0.41	0.38	121±4	28±2	99.14±1.32
Control	101.82±1.27	3.0±0.26	0.27	119±4	33±5	99.58±1.13

\* All values represent mean  $\pm$  standard deviation, n=20; † n=6; ‡ n=3

Copyright to IJARSCT www.ijarsct.co.in



DOI: 10.48175/568





International Journal of Advanced Research in Science, Communication and Technology

International Open-Access, Double-Blind, Peer-Reviewed, Refereed, Multidisciplinary Online Journal

Volume 5, Issue 6, June 2025



#### IN VITRO DISSOLUTION STUDIES OF MCZ FAST DISSOLVING TABLETS



# COMPARISION OF IN VITRO DISSOLUTION STUDIES OF OPTIMIZED MCZ FAST DISSOLVING TABLETS WITH MARKETED TABLETS

The cumulative mean percent of MCZ released from formulation MCZ22 showed more release and found to be  $99.12\pm1.35\%$  in 15 min compared with marketed tablets (Diligan-25).

	Cumulative percent of MCZ released					
	$(Mean \pm S.D.)$					
Time (min)	MCZ22 fast					
	dissolving tablets	Diligan - 25				
0	0.00±0.00	0.00±0.00				
5	44.34±1.58	15.38±1.85				
10	75.26±1.21	37.24±1.61				
15	99.12±1.35	45.71±1.23				
20	-	56.86±1.49				
30	-	67.53±1.57				
45	-	81.62±1.36				
60	-	98.49±1.42				

## STABILITY STUDIES OF OPTIMIZED MCZ FAST DISSOLVING TABLETS

In the stability studies, assay values and drug release were calculated after six months storage The data was subjected to statistical analysis and proved that they were not significantly different from each other (P>0.05).

Copyright to IJARSCT www.ijarsct.co.in



DOI: 10.48175/568





International Journal of Advanced Research in Science, Communication and Technology

International Open-Access, Double-Blind, Peer-Reviewed, Refereed, Multidisciplinary Online Journal

Volume 5, Issue 6, June 2025



Table 7 Stability studies of MCZ22 fast dissolving tablets

Time (min)	Cumulative percent of FLB released		t-test at	5	Factor
	(Mean ± S.D.) Before storage	After 6 months storage	0.05 LS	(F2)	
0	0.00±0.00	0.00±0.00			
5	44.34±1.58	41.67±1.32	Not Significant		
10	75.26±1.21	73.52±1.48		85.74	
15	99.12±1.35	98.27±1.24			
% Assay	99.14±1.17	98.81±1.02	Not Significant		

### **IV. CONCLUSION**

Solid dispersion techniques employing PEGs and Gelucire carriers successfully converted MCZ into an amorphous form, markedly improving its solubility and dissolution behavior. Fast dissolving tablets formulated from these dispersions exhibited rapid disintegration, robust mechanical properties, and significantly enhanced in vitro release compared to a marketed standard. The optimized formulation, MCZ22, achieved nearly complete drug release within 15 minutes and maintained its performance after accelerated stability testing. Consequently, solid dispersion-based FDTs represent a viable strategy to overcome solubility-limited bioavailability of MCZ, offering a promising dosage form for improved patient adherence and therapeutic outcomes

#### REFERENCES

[1] Ahuja, N., Katare, O. P., & Singh, B. (2007). Studies on dissolution enhancement and mathematical modeling of drug release of a poorly water-soluble drug using water-soluble carriers. European Journal of Pharmaceutics and Biopharmaceutics, 65(1), 26–38.

[2] Chiou, W. L., & Riegelman, S. (1971). Pharmaceutical applications of solid dispersion systems. Journal of Pharmaceutical Sciences, 60(9), 1281–1302.

[3] Craig, D. Q. M. (2002). The mechanisms of drug release from solid dispersions in water-soluble polymers. International Journal of Pharmaceutics, 231(2), 131–144.

[4] Vasconcelos, T., Sarmento, B., & Costa, P. (2007). Solid dispersions as strategy to improve oral bioavailability of poor water-soluble drugs. Drug Discovery Today, 12(23–24), 1068–1075.

[5] Leuner, C., &Dressman, J. (2000). Improving drug solubility for oral delivery using solid dispersions. European Journal of Pharmaceutics and Biopharmaceutics, 50(1), 47–60.

[6] Kalepu, S., &Nekkanti, V. (2015). Insoluble drug delivery strategies: review of recent advances and business prospects. ActaPharmaceuticaSinica B, 5(5), 442–453.

[7] Gupta, A. K., &Sahoo, S. (2011). Enhancement of dissolution and bioavailability of poorly soluble drugs by solid dispersion: A review. International Journal of Pharmaceutical Sciences and Research, 2(8), 2021–2030.

[8] Sekiguchi, K., & Obi, N. (1961). Studies on absorption of eutectic mixtures. I. A comparison of the behavior of eutectic mixtures of sulfathiazole and that of ordinary sulfathiazole in man. Chemical & Pharmaceutical Bulletin, 9(11), 866–872.

[9] Chauhan, B., Shimpi, S., &Paradkar, A. (2005). Preparation and evaluation of glibenclamide–polyglycolized glyceride solid dispersions with silicon dioxide by spray drying technique. European Journal of Pharmaceutics and Biopharmaceutics, 61(3), 315–320.

[10] Serajuddin, A. T. M. (1999). Solid dispersion of poorly water-soluble drugs: early promises, subsequent problems, and recent breakthroughs. Journal of Pharmaceutical Sciences, 88(10), 1058–1066.

[11] Jermain, S. V., Brough, C., & Williams, R. O. (2018). Amorphous solid dispersions and nanocrystal technologies for poorly water-soluble drug delivery – an update. International Journal of Pharmaceutics, 535(1-2), 379–392.

Copyright to IJARSCT www.ijarsct.co.in



DOI: 10.48175/568





International Journal of Advanced Research in Science, Communication and Technology

International Open-Access, Double-Blind, Peer-Reviewed, Refereed, Multidisciplinary Online Journal

#### Volume 5, Issue 6, June 2025



[12] Janssens, S., & Van den Mooter, G. (2009). Review: physical chemistry of solid dispersions. Journal of Pharmacy and Pharmacology, 61(12), 1571–1586.

[13] Baghel, S., Cathcart, H., & O'Reilly, N. J. (2016). Polymeric amorphous solid dispersions: A review of amorphization, crystallization, stabilization, and formulation aspects. European Journal of Pharmaceutics and Biopharmaceutics, 101, 31–42.

[14] Patil, J. S., &Paradkar, A. R. (2007). Formulation of a self-emulsifying system for oral delivery of simvastatin: In vitro and in vivo evaluation. ActaPharmaceutica, 57(1), 111–122.

[15] Javadzadeh, Y., Siahi-Shadbad, M. R., Barzegar-Jalali, M., Nokhodchi, A. (2007). Enhancement of dissolution rate of piroxicam using liquisolid compacts. IlFarmaco, 60(4), 361–365





