

Assessment of Physico-Chemical Properties of Coal Mine Water for Sustainable Water Resource Management

Miss. Anjana Chaudhuri¹, Dr. Shikha Shrivastava², Dr. Divya Kumudini Minj³

Government V. Y. T, P.G Autonomous College, Durg, Chhattisgarh^{1,3}

Indira Gandhi PG College, Durg, Chhattisgarh²,

anjana.korba@gmail.com

Abstract: *The aim of our study is to gather various facts from coal mine water regarding seasonal changes, water quality, and abiotic variables. Mine water from coal mines is recycled in the Indian state of Chhattisgarh and used as a thermal energy source as well as in the public sector and on agricultural ground. With the onset of Summer, water samples were obtained for this investigation from coal mine in the Korba district of Chhattisgarh. To evaluate the quality of water, analyses were made of sixteen different physiochemical parameters, such as pH, alkalinity, total hardness, fluorides, chlorides, BOD, COD, and DO. The Water Quality Index (WQI), .. The most useful indicator of the quality of the water is the Water Quality Index (WQI), which is produced by averaging any or all of the features. Therefore, it is safe and there is plenty of source to use mine water as an additional resource for aquatic ecosystems. The preservation and restoration of aquatic ecosystems, as well as the scientific distribution and management of water resources, are essential to the effective use of freshwater. The complete usage and ideal allocation mode of mine water must be taken into consideration based on actual conditions because of the variations in the quantity and quality of coal mine water as well as the notable variations in the surrounding conventional water resource conditions. Additionally, they must include the full utilization of mine water resources into their development plans for the circular economy and local communities.*

Keywords: coal mine water, Monthly variation, Opencast mine .WQI, Physio-chemical parameters

I. INTRODUCTION

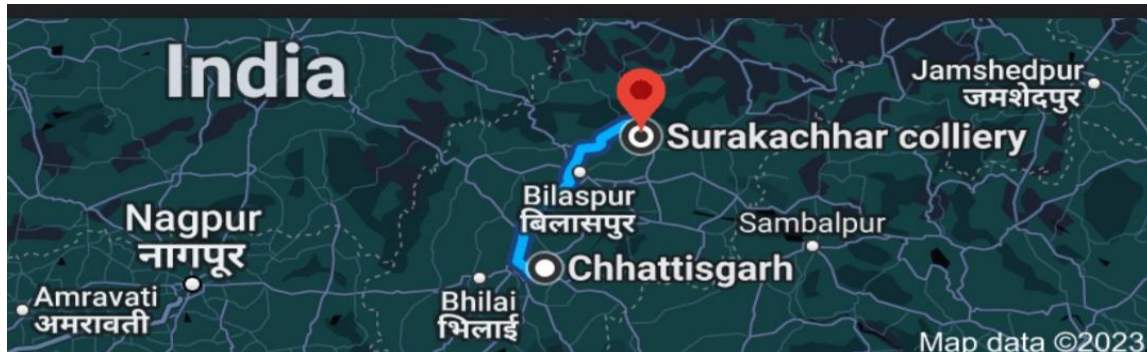
In India, opencast as well as underground coalmines supply almost all of coal. Growing human population, industrialization, and human activity collectively contributed to the extreme pollution of water supplies with various hazardous chemicals. Rock weathering and mining operations naturally pollute water. Owing to the numerous water-borne illnesses, it is very important that the quality of water be tested on a regular basis before it is used for irrigation and house hold work. Obtaining Clean water is essential for reducing illness and enhancing quality of life.. In addition to its quality and quantity, water is a vital component of both urban and rural environments and must be available in appropriate quantities when required in relation to time, place, and environmental factors. As cities and industry expanded, water bodies' levels of contamination rose quickly as well. Mining operations have a significant negative impact on the surface topography and remove vegetation, causing excessive erosion. As one of the most profitable industries in Chhattisgarh is coal mining. In district of Korba, Surakachhar, Dipka, Gevra, and Kusbunda produce the majority of the coal. In this study water sample from underground coalmine of Surakachhar district Korba Chhattisgarh were collected during rainy season, and were analysed to focus water quality of water This study compared the physiochemical characteristics of 6 samples of underground coal mine water with international BIS drinking and household uses, which are based on the Water Quality Index (WQI).Based on weighed arithmetic calculations, Harton established the WQI Model in 1965. Other researchers have also developed a variety of WQI Models (Brownetal1972, GEM, UNEP2007; Kavita and Elongovon 2010; Alobaidy et al. 2010; Sankar and Kavo 2020).The WQI is a numerical measure that has a distinct digital rating system and a value ranking ranging from 0 to 100.It is described as the overall



state of the water quality using the terms bad, good, exceptional, etc., depending on different water quality parameters. Additionally, it is crucial for the management of various water samples to compare them.

MATERIALS AND METHODS

Study Area: The present study area located between latitude (22°.15'N and 22°.30'N) and Longitudes (82°.15'E End 82.55 E) further Korba covers an area of about 530 sq. kilometres. The Surakachhar coal mine is a mine complex operated by South Eastern Coalfields limited .



Climate: The study area comes under -tropical type of climate. It is hot and humid because of it proximity to the tropic of Cancer and its dependence on monsoons for rain. In rainy season begins from June to September sometimes extend upto October, in this area ,climate is sub humid with temperature of 27° C with annual rainfall of 1534 mm.(Himani Kurre,Lata Sharma 2022)

OBJECTIVE:

The causes of frequent decline in the ground water level in Chhattisgarh is due increase in population and insufficient rain fall. Villagers use the river's water, but it gets dried up in Summer. they face lot of difficulties during Summer So,underground mine water is preferred as sopplement source Quality of water directly influence life of colonized population so It is very crucial and significant to assess the water before it is used for drinking, domestic work, agricultural or industrial purpose.. Water may contain different types of floating, dissolved, suspended and microbiological as well as bacteriological impurities.so it is duly tested and the choice of parameters for testing of water, exclusively depends upon for what purpose the water will be used in future and range required for its purity and quality

II. METHODOLOGY

Total 6 numbers of water samples were collected from underground mines during July to September 2024 at morning hours 8.00 AM to 9.00AM The sample bottles were kept air tight and labelled These samples were carried to the laboratory for the analysis of different physicochemical characteristics like pH, total alkalinity, boron, calcium, chloride, colour, fluoride, total hardness, Iron, manganese, nitrate, calcium, turbidity and sulphate The Water quality measured using Water Quality Index WQI under BIS frame work (Arjun Ram, SK Tiwari, Y.V Singh2021,.)The weighed Arithmetic Water Quality Index WQI represented as

$$WQI = \frac{\sum_{i=1}^n W_i Q_i}{\sum W_i} \dots\dots(1)$$

Where n= no.of variable of parameter

W_i = Unit weight for i^{th} parameter

Q_i = Quality rating sub index of i^{th} water quantity parameter

W_i are inversely proportional to the recommended standard for the corresponding parameters

$W_i = K/S_n \dots\dots(2)$ S= Standard value for i^{th} parameter.



K= Proportional Constant

The value of $K = 1 / \sum 1/S_n \dots \dots \dots (3)$

The value of $QI = Q / 100 [(V_o - V_i) / (S_n - V_i)] \dots \dots (4)$

Where V_o = Observed value of i^{th} parameter

V_i = Ideal value of i^{th} parameter in pure water V_i are taken as 0 for drinking water except for pH and dissolved Oxygen(DO) *Tripathy et al 2005*).

For pH ideal value is 7.0 and permissible value is 7.0. Similarly for DO ideal value is 14.6 mg/l and permissible value is 5.0 mg/l. So the quantity rating for pH and DO are calculated from the equation respectively shown below.

$Q_{pH} = 100$ $Q_{DO} = 100$

$[V_{pH} - 7.0] / 8.5 - 7.0$ $[V_{DO} - 14.6 / 5.0 - 14.6] \dots \dots \dots (5)$

Where V_{pH} = Observed value of (DO).

Classification of water quality based on WQI Chatterjee and Raziuddin (2002).

SNO	WQI	RATING CLASS
1	0-25	EXCELLENT
2	26-50	GOOD
3	51-75	POOR
4	76-100	VERY POOR
5	>100	UNSUITABLE

Table 1. Water quality index of mine water July to September

VALUE OF WATER QUALITY INDEX OF SIX UNDERGROUND MINE WATER SAMPLES									
SAMPLE=S1 TO S6				JULY		AUGUST		SEPTEMBER	
			BIS	s1	s2	s3	s4	s5	s6
S.No	PARAMETER	METHOD							
1	PH	Potentiometer	8.5	7.35	6.54	8.43	8.5	8.45	8.1
2	Turbidity	Nephelometric Turbidity	5	8.16	14	14.5	14.6	20	19
3	Total Hardness	Complexometry EDTA Titration	200	198.84	154	151	151	154	153
4	TDS	Gravimetric method	500	406	557	407	402	405	407
5	Chloride	Argentometric titration	250	77.43	75.8	76	77	77.3	77.1
6	Alkalinity	Argentometric titration	200	223	224	200	201	125	121
7	Calcium	Argentometric titration	7.5	36	47.98	36	34	35	29



8	Magnesium	Argentometric titration	30	15.3	19.23	15.3	15.2	13.4	13.4
9	Iron	Spectrometer	0.3	0.3	0.3	0.3	0.31	0.34	0.35
10	Nitrate	Spectrometer	45	0.2	0.3	0.4	0.54	0.34	0.38
11	Sulphate	Spectrometer	200	38	23	32	31	30	35
12	Fluoride	ion selective electrode	1	0.26	0.25	0.23	0.24	0.26	0.25
13	BOD PPM	Incubation method	10	10	19	10	16	21	18
14	CODmg/l	Titration method	100	70	56	70	81	75	76
15	DO mg/l	Titration method	5	12	11	12	10	10	11
16	EC(S/m)	Conductivity Probe	300	231	240	231	232	234	230
WATER QUALITY INDEX				93.44	102.01	97.77	99.55	111	104.89

Table 2. Value of water quality index of six underground mine water samples

III. RESULT AND DISCUSION

During the summer, groundwater and river water are insufficient to meet water demands, making clean water essential for a healthy living. So water of rainy season must be stored and utilized .From July to September 2024, six underground mine water samples were tested for physicochemical characteristics such as pH, color, turbidity, hardness, chloride, alkalinity, and dissolved particles, and compared to BIS drinking water quality standards.The Water Quality Index is also computed.This study is based on selected parameters and BIS considered as standard for reference.

pH –It is one of important indicator for assignment of water.in present study it ranges from 6.54 to8.45 ideal for consumption(.fig1)

Turbidity: Water turbidity consists from suspense inorganic substances, dispersion organic substances, micro microorganisms etc. Turbidity is important because it affects both the acceptability of water to consumers, and the selection and efficiency of treatment processes, particularly the efficiency of disinfection with Chlorine since it exerts a Chlorine demand and protects microorganisms and may also stimulate the growth of bacteria. The turbidity of water samples varied from 18.6 to 19 NTU (Average: 3.7 NTU). So in the present study it is higher value I not fit for domestic use.(fig .2)

Alkanity: The highest alkalinity level was measured in July and the lowest in September The alkalinity of the water reached its peak due to an increase in bicarbonates. Three of the six water samples had alkalinity readings that were within the acceptable range.(Fig3).

Calcium: The presence of Calcium in samples is directly proportional to the hardness of water. The calcium concentrations in the samples ranged from 29 to 47.38 mg/L.Calcium levels peaked in February and fell to their lowest point in April. Except for the month, the calcium concentration of all water samples was within the acceptable range.(Fig4)

Chloride: : Chloride is often associated with sodium since sodium chloride is a common constituent if some water sources, the levels above 140 mg/L are considered to be toxic for plants (Flood, 1996). The chloride contents indicate domestic as well as industrial pollution (*Chatterjee, et al., 2002*). The values of chlorides in different months were found to be various ranges from 75.8 mg/L to 77.43 mg/L. One of the most crucial factors in determining the quality of water is chloride.According to Munawar (1970), higher chloride concentrations indicate more organic contamination. The chloride concentration of all of the water samples was generally within the required range.



Fluoride: The fluoride concentration of water samples ranged from 0.23 mg/l to 0.26 mg/l. The lowest value of fluoride was found in the month of March, while the highest was recorded in the months of February and March. Overall, the fluoride concentration of all water samples was below acceptable limits. Total hardness is a characteristic of water that causes leather development with soap and increases the boiling point of water.

Sulphate: In the present study it ranges from 23mg/L to 38mg/L. All samples have Sulphate value within permeable range which is deal for domestic purpose.

Hardness : Hardness of water mainly depends upon the amount of magnesium and calcium salts dissolved (Trivedi and Goel, 1986). The value of total hardness of water samples in the present study fluctuated from 150mg/L to 198.84mg/L. 90% of the water samples were showing under the desirable limit.

Iron: Iron is the important metal for life of vegetable and animal organisms. It is undesirable for household and for industry. Iron concentration in all the samples was fluctuating from 0.3mg/L to 0.34 mg/L. Generally, the concentration of iron of all the samples was within the desirable limit.

Magnesium: It is often associated with calcium in all kinds of water, but its concentration remains generally lower than the calcium. In the present investigation, the results of Magnesium of water samples in all the months were vary from 13.4mg/L to 19.3mg/L.

Nitrate: Nitrate is present final products of biological oxidation form organic pollution. Nitrate concentration depends on the activity of nitrifying bacteria which in turn get influenced by presence of dissolved oxygen. This signifies that in the most time water where polluted. The values of nitrates in the study area in different months were varied from 0.2 mg/L to 0.54mg/L. The lowest and highest value was measured in February and April respectively (High concentrations of nitrates increase the growth of vegetation in water systems and elevate oxygen demand (Mc Junkin, 1982). Generally, the nitrate content of all the water samples was within limit.

TDS: Dissolved solids value of water samples in all the months were found to be exceed the desirable to maximum Dissolved Solids. In natural waters, dissolved solids consists mainly of inorganic salts such as carbonates, bicarbonates, chlorides, sulphates, phosphates and nitrates of calcium, magnesium, sodium, potassium, iron etc. and small amount of organic matter and dissolved gases. The values of dissolved solids of the water samples ranged 10mg/L to 12 mg/L.

BOD: ranges from 56mg/L to 81mg/L is enumerating the amount of oxidizable pollutant found in underground water and BOD for healthy drinking water 1PPM to 2PPM. In present study it ranges 10PPM to 19PPM from, not fit for consumption.

Electrical Conductivity: The value of EC ranges from 230 S/m to 241 S/m. All the samples have EC value within the limit. Approximately six physiochemical parameters were used to assess the water quality of coal mine water. Parameters include pH, alkalinity, total hardness, chlorides, fluorides, BOD, COD, DO, nitrate, sulphate, TDS, and electrical conductivity, among others.

IV. CONCLUSION

All of the water samples were examined, and it was discovered that the water quality is poor, with the exception of turbidity, calcium, fluoride, and total hardness being slightly higher than the acceptable value. As a result, it is safe for human health. As a result, there is abundant opportunity to use mine water for drinking and household reasons to meet local demands after some treatment. To regulate water pollution, the industry, State Pollution regulate Board, and government must take proper efforts to prevent water pollution. The mentioned preventive steps can be very useful. During the rainy season groundwater and river water are sufficient to meet water demands, and clean water is vital for a healthy life. The current study shows the Water Quality Index of a sample collected from a coal mine in Surakachhar is unsuitable for drinking but there are several opportunities to use mine water for domestic and drinking purposes following treatment.

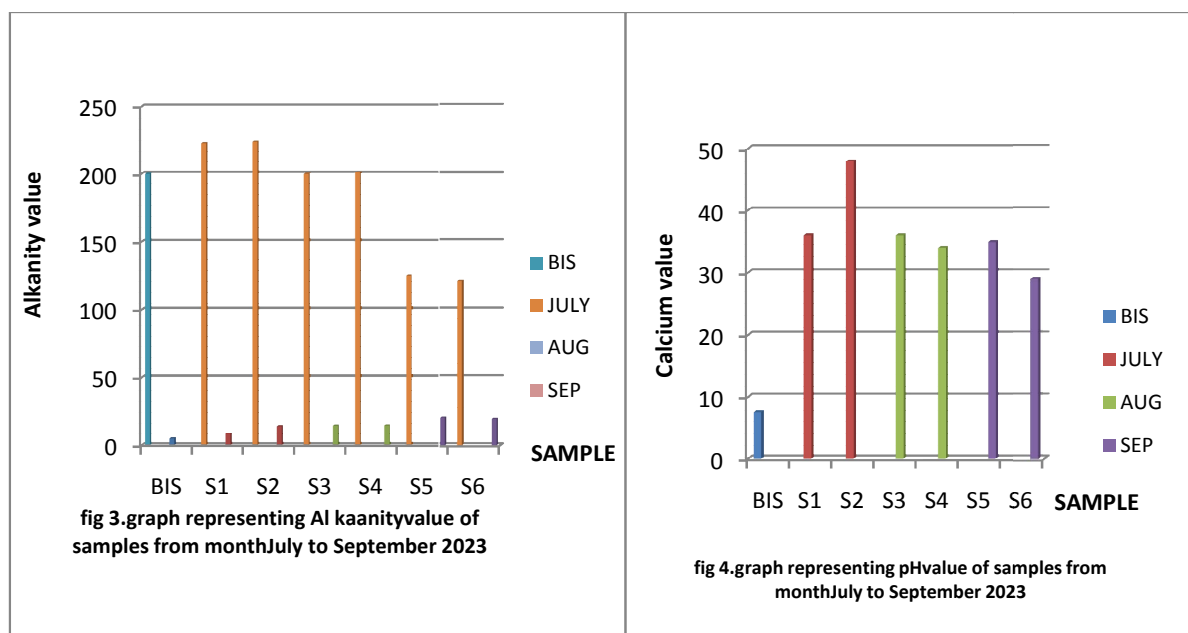
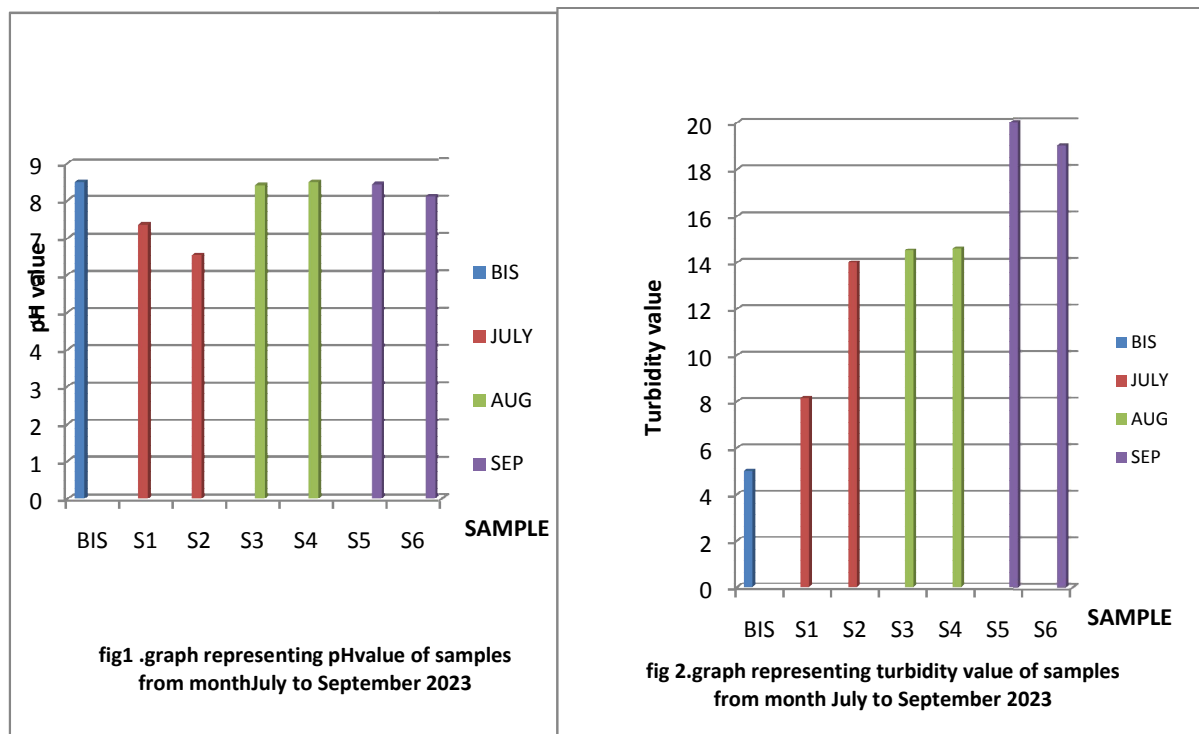
ACKNOWLEDGMENT

Our thankful acknowledge to Principal, Government V. Y. T Autonomous PG, college, Durg, Chhattisgarh, Manager and mine workers of Surakachhar coal mine and Department of Zoology VYT Autonomous PG, Durg, Chhattisgarh, Agrasen Girls College Korba, C.G for providing us necessary laboratory facilities for the present work.



CONFLICT OF INTEREST:

The authors declare that there is no conflict of interests regarding the publication of this paper



REFERENCES

- [1]. APHA(1998), Standard methods for the examination of water and waste water. 20th Ed., American public health Association, Washington, 1998.
- [2]. Aravindan ,S., Shankar, K., Ganesh, B.P., Rajan, K.D. 2010. Hydro geochemical mapping of in the hard rock area of Gadilam River basin, using GIS technique. Tamil Nadu Indian J Appl Geochem 12(2):209–216
- [3]. Balamurugan. P., Kumar, P.S., Shankar, K. 2020. Dataset on the suitability of groundwater for drinking and irrigation purposes in the Sarabanga River region, Tamil Nadu, India. Data in Brief 29:105255
- [4]. Bhatnagar A, Singh G. 2010. Culture fisheries in village ponds: a multi-location study in Haryana, India. Agriculture and Biology Journal of North America. 1(5):961-968.
- [5]. Bhatnagar A. 2008, Productivity and fish biodiversity of selected ponds of Haryana, Project Report submitted to Department of fisheries Government of Haryana.
- [6]. Bhatnagar, A., Garg, S.K.2000. Causative factors of fish mortality in still water fish ponds under sub-tropical conditions, Aquaculture.; 1(2):91-96 Solis NB. The Biology and Culture of Penaeus Monodon, Department Papers. SEAFDEC Aquaculture Department, Tigbouan, Bolo Philippines, 1988, 3-36.
- [7]. Bhatnagar, A., Jana, S.N., Garg, S.K., Patra, B.C., Singh, G.,Barman,U.K. 2004. Water quality management in aquaculture, In: Course Manual of summer school on development of sustainable aquaculture technology in fresh and saline waters, CCS Haryana Agricultural, Hisar ,India203-210.
- [8]. BIS, Indian standard specification for drinking water. IS: 10500, Indian Standard Institute, 1991.
- [9]. Boyd, S.R., Hall, A., and Pillinger, C.T. 1993. The measurement of d15N in crustal rocks by static vacuum mass spectrometry: Application to the origin of the ammonium in the Cornubian batholith, southwest England. Geochim Cosmochim Acta.
- [10]. Brhane. G.K. 2018. Characterization of hydro chemistry and groundwater quality evaluation for drinking purpose in Adigrat area, Tigray, northern Ethiopia. Water Sci 32:213.
- [11]. Brown ,R.M., Mclelland, N.J., Deiniger ,R.A., O'Connor, M.F.1972. Water quality index-crossing the physical barrier. Res Jerusalem 6:787–797.
- [12]. Carlos, D. and James, L. 1997. Golden Dreams, Poisoned streams.Washington D.C. Mineral Policy Center.
- [13]. Carrillo-Rivera, J.J., Cardona, A., Edmunds, W.M.,2002 .Use of abstraction regime and knowledge of hydrogeological conditions to control high-fluoride concentration in abstracted groundwater: San Luis Potosi basin, Mexico. J Hydrol 261:24–47.
- [14]. Chatterjee C, Raziuddin M (2002) Determination f Water Quality Index (WQI) of a degraded river in Asansol industrial area (West Bengal). Nat Environ Pollutes Techno 1:181–189.
- [15]. Chatterjee, C. and Raziuddin, M. 2002. Determination of Water Quality Index (WQI) of a degraded river in Asnol Industrial Area, West Bengal.Jour. of Env and Poll. 1 (2) : 181-189.
- [16]. Chaurasia AK, Pandey HK, Tiwari SK, Prakash R, Pandey P, Ram A (2018) Groundwater quality assessment using water quality index (WQI) in parts of Varanasi district, Uttar Pradesh, India. J Geol Soc India 92:76–82
- [17]. Deshmukh, A.N., Shah, K.C., Sriram, A. 1995 Coal Ash: a source of fluoride pollution, a case study of Koradi thermal power station, District Nagpur. Maharashtra Gondwana Geol Mag 9:21–29Diersing N, Nancy F (2009) Water quality: Frequently asked questions. Florida Brooks National Marine Sanctuary, Key West, FLK (2010) Ground water quality characteristics at Erode district, Tamilnadu India. Int J Environ Sci 1:163–175.
- [18]. Dhar, B.B. and Rolterdem, 1993.Environment Management and Pollution Control in Mining Industry. APH, New Delhi.
- [19]. Ekubo, A.A., Abowei, J.F.N. 2011. Review of some water quality management principles in culture fisheries, Research Journal of Applied Sciences, Engineering and Technology; 3(2):1342-1357.
- [20]. Ellis, K.V. 1989. Surface Water Pollution and its Control. Macmillan Press Ltd, Hound Mill, Basingstoke, 3-18.



- [21]. Flood, D. 1996. Irrigation Water Quality for BC Greenhouses, Floriculture Fact sheet, Ministry of Agriculture, Fisheries and Food, British Columbia. Gupta, P.K. 2004. Soil, Plant, Water and Fertilizer Analysis. Shyam Printing Press, Agrobios, India, pp 438
- [22]. G.K. Nigam, G.K. Sahu, R.K. Sinha, J., Sonwani, R.N. 2015. A study on physio-chemical characteristics of open cast mine Jr. of Industrial Pollution Control 31(2)(2015) pp 191-200.
- [23]. Garg SK, Bhatnagar A. 1999. Effect of different doses of organic fertilizer (cow dung) on pond productivity and fish biomass in still water ponds. Journal of Applied Ichthyology, 15:10-18.
- [24]. Garg, S.K., Bhatnagar, A. 1996. Effect of varying doses of organic and inorganic fertilizers on plankton production and fish biomass in brackish water ponds, Aquaculture Research (The Netherlands). 27:157-166.
- [25]. Hari, O.S., Nepal, M.S. Aryo and Singh, N. 1994. Combined effect of waste of distillery and sugar mill on
- [26]. Karanth, K.R. 1987. Groundwater Assessment Development and Management. Tata McGraw Hill Publishing Company Ltd., New Delhi: 725.
- [27]. Himani Kurre Sandey, Lata Sharma. Phytochemistry and Ethanomedical leafy plant used by tribals people of District Kondagaon Chhattisgarh. International journal of advanced research in biotechnology and nanobiotechnology. Vol3, Issue I, July 2022.
- [28]. Kawo, N.S, Shankar, K. 2018. Groundwater quality assessment using water quality index and GIS technique in Modjo River Basin, Central Ethiopia. J Afr Earth Sci 147:300–311
- [29]. Kiran, B.R. 2010. Physico-chemical characteristics of fish ponds of Bhadra project at Karnataka, RASĀYAN Journal of Chemistry, 3(4):671-676.
- [30]. Kulkarni, G.J. 1997. Water Supply and Sanitary Engineering. 10th Ed. Farooq Kitabs Ghar. Karachi, 497.
- [31]. Mack N. Pond water chemistry, San Diego, Koi Club, 1996; [Http://users.vcnet.com/rrenshaw/h2oquality.html](http://users.vcnet.com/rrenshaw/h2oquality.html) Revised on. 1996
- [32]. Majumdar, D., Gupta, N. 2000. Nitrate pollution of groundwater and associated human health disorders. Indian J Environ Health 42:28–39.
- [33]. McJunkin, F.E. 1982. Water and Human Health. United States Agency for International Development, Washington, DC.
- [34]. Mumtazuddin, M., Rahman, M.S., Mostafa, G. 1982. Immunological studies of four selected rearing ponds at the aquaculture experiment station, Mymensingh. Bangladesh Journal of Fisheries Research. 2-5(1-2):83-90.
- [35]. Munawar, M. 1970. Limnological Studies on Freshwater Ponds of Hyderabad, India. The Biotope, Journal Hydrobiologia. 35: 127-162.
- [36]. Naganandini, M.N., Hosmani, S.P. 1998. Ecology of certain inland waters of Mysore district, occurrence of Cyanophycean bloom at Hosakere Lake, Pollut. Res. 17(2):123-125.
- [37]. Pandey, J, Pandey, Usha, Tyagi, H.R. 2000. Nutrient Status and Cyanobacterial diversity of a tropical fresh water lake. J. Environ. Biol. 21(2):133-138.
- [38]. Patil DB, Tigre RV. 2001. Studies on Water quality of Godchiroli Lake. Poll. Res.; 20:257-259.
- [39]. Peavy Howards, S., Rowe, D.R. and Tehobanoglous, G. 1986. Environmental Engineering. Pub. McGraw-Hill Int. Edition. Civil Engineering Series..
- [40]. Radha Krishnan, R., Dharma raj, K. and Ranjitha Kumari, B.D. (2007). A comparative study on the physico-chemical and bacterial analysis of drinking bore well and sewage water in the three different places of Sivakasi. J. Environ. Biol. (28) : 105-108.
- [41]. Rana, N., Verma, M., Jain, S. 2016. Assessment of different Water Quality Parameters Of Water Sources Of Meerut Region (Uttar Pradesh, India) For Suitability Of Fish Production. Journal of Environmental Bio-Sciences., 30(2):0973-6913 (Print), ISSN 0976-3384 (On Line).
- [42]. Sachidanandamurthy, K.L. and Yajurvedi, H.N. (2006). A study of physico-chemical parameters of an aquaculture body in Mysore city, Karnataka, India. J. Environ. Biol. 27 : 615-618.



- [43]. Sadashivaiah, C., Ramakrishnaiah, C.R., Ranganna, G. 2008. Hydrochemical Analysis and Evaluation of Groundwater Quality in Tumkur Taluk, Karnataka State, India, *International Journal of Environmental Research Public Health*; 5(3):158-164.
- [44]. Scharer, J.M., Pettit, C.M., Kirkaldy, J.L., Bolduc, L., Halbert, B.E. and Chambers, D.B. (2000). Leaching of metals from sulphide mine wastes at neutral pH. In: *Proceedings from the Fifth International Conference on Acid Rock Drainage, ICARD 2000*. Society for Mining, Metallurgy, and Exploration, Inc., 191-201
- [45]. seed germination, seedling growth and biomass of okra. *Journal of Environmental Biology*. 3 (15) : 171-175.
- [46]. Simpi, B., Hiremath, S.M., Murthy, K.N.S., Chandrashekarappa, K.N., Patel, A.N. and Puttiah, E.T. 2011. Analysis of water quality using physicochemical parameters Hosahalli tank in Shimoga District, Karnataka, India.
- [47]. Stiassny M. 1999. The Medium is the Message: Freshwater Biodiversity in Peril. In *the living Planet in Crisis: Biodiversity science and policy*. Columbia University Press, New York,
- [48]. Stumm, W., Morgan, J.J. 1981. An introduction emphasizing chemical equilibria in natural waters, *Aquatic chemistry*. 2ND Ed., John Wiley and Sons, New York, , 780.
- [49]. Swann, L.D. 2004. A Fish Farmer's Guide to Understanding Water Quality, *Aquaculture Extension Illinois*, Purdue University, Indiana Sea Grant Program Fact Sheet AS-1997, 503, . of Arkansas at Pine Bluff Aquaculture / Fisheries
- [50]. Tambekar, P., Morey, P., Batra, R. J. and Weginwar, R. G., 2013. Assessment of water on upstream and downstream from HLOCCM, of Chandrapur Tehsil of Chandrapur District (M.S.). *Jr. of Chemical and Pharmaceutical Research*. 5 (5) : 18-26.
- [51]. Tiwari, A.R., Pandey, S.K., Chaurasia, H.P., Singh, S., Singh, Y.V. 2021. Groundwater quality assessment using water quality index (WQI) under GIS framework, [Applied Water Science](#) volume 11, Article number: 46
- [52]. Trinova, I.S. 1989. Change in community structure productivity of phytoplankton as indicator of lake reservoir eutrophication, *Archiv für Hydrobiologie–Beiheft Ergebnisse der Limnologie*, 33:363-371.
- [53]. Trivedi, R.N., Dubey, D.P. and Bharti, S.L. 2009. Hydrogeochemistry and groundwater quality in Beehar river basin, Rewa District, Madhya Pradesh, India, *Proc. International Conference on Hydrology and Watershed*, JN & T Hyderabad: 49-59.
- [54]. Trivedy, R.K. and Goel, P.K. 1986. *Chemical and Biological Methods for Water Pollution Studies*, Environmental Publication, Karad, pp: 217.
- [55]. WHO. World Health Organization. *Guidelines for Drinking Water Quality*, World Health Organization, Geneva, Switzerland, 1993.
- [56]. Williams, E.G., Rose, A.W., Parizek, R.R. and Water, S.A. 1982. *Factors Controlling the Generation of Acid Mine Drainage*. Final Report on U. S Bureau of Mine Research. Grant No GS105086, University Park, Pennsylvania State University, pp 265.
- [57]. World Resources Institute, 2000. *World Resources, 1999-2000*. Basic Books Inc. New York

